

Lecture 8 CH101 A2 (MWF 11:15 am) Fall 2017 Copyright © 2017 Dan Dill dan@bu.edu

[TP] Express 2000 cm^{-1} in m^{-1} .

20% 1. 2000 m^{-1}
 20% 2. 200000 m^{-1}
 20% 3. 0.05 m^{-1}
 20% 4. 0.0005 m^{-1}
 20% 5. Something else

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Lecture 8 CH101 A2 (MWF 11:15 am)
 Friday, September 22, 2017

For today ...

- Review: Jiggling of bonded atoms
- Wavelength, frequency, and wavenumber


Next lecture: Infrared (IR) spectra; Ch4: (secs 1, 3, and 4 only):
 Atmospheric warming
 Think about It e4.4: Collisional heating <http://goo.gl/vQ0Nz>
 Think about It e4.5: IR windows <http://goo.gl/I8IGz>

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Jiggling of bonded atoms

Lighter faster; stronger faster; dissimilar approaches lighter
 Interactive exploration
<http://quantum.bu.edu/CDF/101/IRFrequency.cdf>

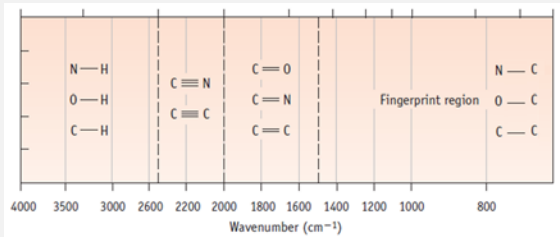


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IR spectra detect function groups

Lighter faster; stronger faster; dissimilar approaches lighter




Wavenumber (cm^{-1})

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Wavelength, frequency, and wavenumber

6



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What's "nu"?

Frequency $\nu = c/\lambda$


Speed of light $c = 299792458 \text{ m/s}$ ($\approx 3 \times 10^8 \text{ m/s}$)

Energy exchanged with matter $\propto \nu$

Since $1/\lambda \propto \nu$, ...

energy exchanged with matter also \propto wavenumber $1/\lambda$

8



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What's "nu"? $\nu = c/\lambda$, $c = 299792458 \text{ m/s}$

Get practice by working through [tutorials e3.15](#) (p 75)

What is the frequency of orange light, $\lambda = 600 \text{ nm}$?

Answer: $c/\lambda = 5 \times 10^{14}/\text{s}$

9



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What's "nu"? $\nu = c/\lambda$, $c = 299792458 \text{ m/s}$

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Wavelength (in nm) and visibility of $2 \times 10^{15}/\text{s}$ light?

Answer: $\lambda = c/(2.0 \times 10^{15}/\text{s}) = 150 \text{ nm}$; in UV so **not visible**

10



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wavelength λ , frequency ν , wavenumber $\tilde{\nu} = 1 / \lambda$

Wavenumber is the number of waves (wavelengths) that fit in one cm.

Here is figure spanning 1 cm, showing a wave with speed $u = \nu\lambda = 1 \text{ cm/s}$.

So, $\nu = 3/\text{s}$ and $\lambda = 1/3 \text{ cm}$.

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11

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wavelength λ , frequency ν , wavenumber $\tilde{\nu} = 1 / \lambda$

Wavenumber is the number of waves (wavelengths) that fit in one cm.

The figures shows that there are 3 wavelengths in 1 cm.

So, in this case the wavenumber is $\tilde{\nu} = 3/\text{cm}$.

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12

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wavelength λ , frequency ν , wavenumber $\tilde{\nu} = 1 / \lambda$

Wavenumber is the number of waves (wavelengths) that fit in one cm.

This figure shows that there are 7 wavelengths in 1 cm.

So, in this case the wavenumber is $\tilde{\nu} = 7/\text{cm}$.

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13

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wavelength λ , frequency ν , wavenumber $\tilde{\nu} = 1 / \lambda$

Wavenumber is the number of waves (wavelengths) that fit in one cm.

Larger wavenumber means smaller wavelength.

$\tilde{\nu} = 3/\text{cm}, \lambda = 1/3 \text{ cm}$ $\tilde{\nu} = 7/\text{cm}, \lambda = 1/7 \text{ cm}$

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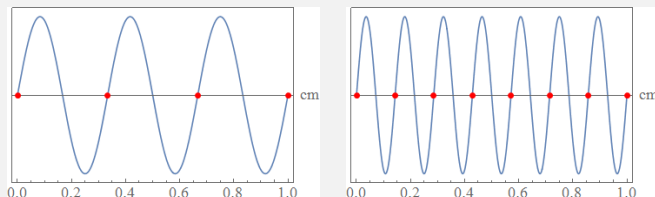
14

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wavelength λ , frequency ν , wavenumber $\tilde{\nu} = 1 / \lambda$

Wavenumber is the number of waves (wavelengths) that fit in one cm.

Larger wavenumber means larger frequency.



$\tilde{\nu} = 3/\text{cm}, \nu = 3/\text{s}$ $\tilde{\nu} = 7/\text{cm}, \nu = 7/\text{s}$

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15

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[TP] Express $2000 \text{ cm}^{-1} = 2000/\text{cm}$ in m^{-1} .

20% 1. $2000 \text{ m}^{-1} = 2000/\text{m}$
 20% 2. $200000 \text{ m}^{-1} = 200000/\text{m}$
 20% 3. $0.05 \text{ m}^{-1} = 0.05/\text{m}$
 20% 4. $0.0005 \text{ m}^{-1} = 0.0005/\text{m}$
 20% 5. Something else

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16

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wavelength λ , frequency ν , wavenumber $\tilde{\nu} = 1 / \lambda$

The frequency of a light wave (speed = c) with $\tilde{\nu} = 2000 \text{ cm}^{-1} = 2000/\text{cm}$ is ...

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17

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wavelength λ , frequency ν , wavenumber $\tilde{\nu} = 1 / \lambda$

The wavelength of a light wave (speed = c) with $\tilde{\nu} = 2000 \text{ cm}^{-1} = 2000/\text{cm}$ is ...

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18


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What's "nu"? $\nu = c / \lambda$, $c = 299792458 \text{ m/s}$

Get practice by working through [tutorials e3.15](#) (p 75)

What is frequency of motion at $\tilde{\nu} = 3000 \text{ cm}^{-1}$?

Answer: $c \times 1/\lambda = 9 \times 10^{13}/\text{s}$; in IR so **not visible**

 19

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wavelength λ versus wavenumber $\tilde{\nu} = 1 / \lambda$

Given wavelength λ , get frequency ν by **dividing it into** c .

$\lambda = 3000 \text{ nm} \rightarrow \nu = c / \lambda =$
 $\nu = (3 \times 10^8 \text{ m/s}) / (3000 \times 10^{-9} \text{ m}) = 1 \times 10^{14}/\text{s}$

Given wavenumber $\tilde{\nu} = 1 / \lambda$, get ν by **multiplying it by** c .

$\tilde{\nu} = 3000 \text{ cm}^{-1} \rightarrow \nu = c \times \tilde{\nu} =$
 $\nu = (3 \times 10^{10} \text{ cm/s}) \times (3000 \text{ cm}^{-1}) = 9 \times 10^{13}/\text{s}$

 20

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[Quiz] Which has the **lowest** frequency (of jiggling)?

0% 1. An antisymmetric H stretch at $9 \times 10^{13}/\text{s}$

0% 2. A sulfide, C-S, stretch at 600 cm^{-1}

0% 3. Red light at 700 nm

0% 4. Visible light at $6 \times 10^{14}/\text{s}$

 21