1. Cu$_2$+|Cu = +0.34 V, H+|H$_2$ = 0 V; Zn$_2$+|Zn = -0.76 V
   Placing Cu(s) in 0.2 M HCl(aq), H$_2$ bubbles ...
   
   A. should form
   B. should not form
   C. further information required

2. Cu$_2$+|Cu = +0.34 V, H+|H$_2$ = 0 V; Zn$_2$+|Zn = -0.76 V
   Placing Zn(s) in 0.2 M HCl(aq), H$_2$ bubbles ...
   
   A. should form
   B. should not form
   C. further information required

3. Cu$_2$+|Cu = +0.34 V, H+|H$_2$ = 0 V; Zn$_2$+|Zn = -0.76 V
   Placing Zn(s) in CuSO$_4$(aq), ...
   
   A. H$_2$ bubbles should form
   B. no reaction should occur
   C. Cu(s) should deposit on the Zn(s)
   D. further information required
   E. 

4. Cu$_2$+|Cu = +0.34 V, H+|H$_2$ = 0 V; Zn$_2$+|Zn = -0.76 V
   Placing Zn(s) coated with Cu(s) in 0.2 M HCl(aq), compared to H$_2$ bubble formation with pure Zn, H$_2$ bubbles should appear ...
   
   A. more slowly
   B. at the same rate
   C. more quickly